## Inactivation of HF in Li-Ion Batteries, EDLCs and Supercapacitors Enhanced lifetime and safety through functional glass powders

Hydrofluoric acid (HF) is responsible for degradation and a shortened lifetime in numerous energy storage devices that use fluorine-containing organic electrolytes, e.g. lithium-ion batteries. Even small amounts of residual moisture (e.g. introduced during battery production) often result in the formation of highly corrosive HF inside the battery cells, since flourine reacts violently with water. This affects the electrolyte chemistry, leading to lower performance, failure of the device, or in worst case outgassing and leakage of the electrolyte.

### **Product information**

SCHOTT has developed a special Ba-silicate glass powder to absorb hydrofluoric acid. Due to its special composition, the inorganic glass particles can scavenge fluorine ions and chemically bind them to the particle surface. This binding is permanently inactivating and preventing the formation of hydrofluoric acid in the battery system.

## Advantages of SCHOTT HF-scavenging glass powder

- Significant increase of lifetime and safety of lithium-ion batteries, EDLCs and Supercaps (dependent on materials/chemicals used)
- In case of thermal run-away, the glass powder particles remain in place and protect the electrodes from contacting each other, preventing a short circuit reaction when used on or in the separator.
- The world's first inorganic additive with HF-gettering function:
  - Chemically resistant to fluor containing electrolytes
  - Electrochemically stable even in high voltage applications
  - Temperature resistant up to several hundreds °C
  - Reduced abrasion in coating processes due to lower hardness in comparison to ceramic materials

Li-Ion Electrolyte				
Electrolyte	Step 1 $Li^+$ $PF_6^-$ $Li^+$ $H_2O + PF_6^ H_{F^-}^+$	Step 2   Li* PF <sub>6</sub> <sup>-</sup> PF <sub>6</sub> <sup>-</sup> Li*   PF <sub>6</sub> <sup>-</sup> Li*	Step 3 $PF_6^- PF_6^-$ Li* Li* $PF_6^-$ Li* HF-Complex	
Glass Surface Glass bulk	OHOHOHOH - Si - O   Ba   O - Si -   O O O   - Si - O - Si - O - Ba -   - Si - O - Si - O - Ba -	OHOHOHOHOHOHOHOH_	OH - Si - OI - SI -	

SiO<sub>2</sub>-based glass powder can scavenge fluorine ions and chemically bind them to the glass surface





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#### **Material Data**

	HF Scavenging Glass G018-405		
Form of delivery/particle size	Glass powder with a mean particle size of 1µm or 0.7 µm (d50) and 90% of particles smaller than 5 µm (d90). Other grain sizes are available upon request.		
Approx. specific surface area	m²/g		
HF scavenging power	mg F <sup>.</sup> / g glass	40 – 60	
Coefficient of thermal expansion (-30 / +70 °C)	10 <sup>.6</sup> /K	10.1	
Max. operation temperature (ISO 7884-8)	°C	> 600	
Density	g/cm <sup>3</sup>	4.1	
Young's Modulus	GPa	68	
Hardness (Knoop) HK0.1/20	-	440	
Wettability with electrolyte (contact angle bulk)	°C	15-30	
Residual water (Karl Fischer after drying 100 °C/4h/air)	mg/cm <sup>3</sup>	6	

## **Applications**

SCHOTT's HF-gettering glass powder is suitable as an additive for the separator, the electrode materials, as a coating on electrode materials, or in the electrolyte in:

- Lithium-ion batteries
- Electric double layer capacitors (EDLC) or Supercapacitors
- It can also provide benefits for other applications, in which HF must be chemically bound

